



Modification de surface par procédés plasmas

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Brief History of Plasma Physics...

1879

W. Crookes defines the state of a ionized gas as "... a world where matter may exist in a 4th state ...".

1927

Irving Langmuir first used this term to describe an ionized gas in 1927--Langmuir was reminded of the way blood plasma carries red and white corpuscles by the way an electrified fluid carries electrons and ions

Irving Langmuir (1881-1957)



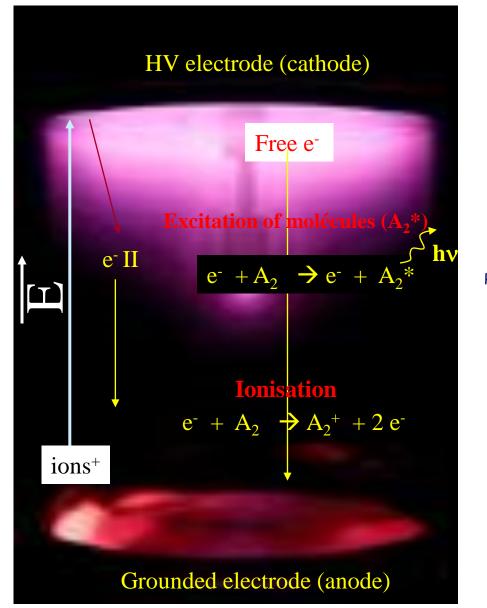
Nobel Laureate in Chemistry in 1932... for his discoveries and investigations in surface chemistry ...

Definition of Plasma ...



Langmuir uses the word Plasma to define a neutral partially or fully ionized gas, containing neutral (atoms, molecules, radicals) and charged (ions, electrons) species.

GENERATION OF A PLASMA



AN ELECTRIC FIELD IS IMPOSED TO A GAS



COLLISIONS AND ENERGY EXCHANGE PROCESSES OCCUR, MORE IONS AND ELECTRONS ARE FORMED, THE IONIZATION DEGREE INCREASES, MOLECULES ARE FRAGMENTED, HOMOGENEOUS AND HETEROGENEOUS REACTIONS OCCUR

THE PLASMA IS SUSTAINED BY A BALANCE BETWEEN PRODUCTION (ionizations) AND LOSS (recombinations) OF CHARGED SPECIES

> Reactive species : Ions, electrons, radicals, Excited species, photons...

Three classes of plasmas...

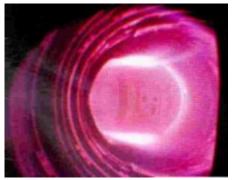
<u>« Hot plasmas »</u>

Millions of degrees

Sun



Tokamak



→ITER: International Thermonuclear Experimental Reactor <u>« Thermal plasmas »</u> Few thousands of degrees





Welding, Metallurgy, Plasma spray deposition, ICP spectroscopy, Waste abatment

<u>« Cold plasma »</u>

Few hundreds of degrees





→Surface treatment of thermosensitive materials

Brief History of Plasma processing...

70's

•First plasma etching processes

•Plasma processes started to be applied for surface modification of materials, in the field of microelectronics and semiconductors

80's

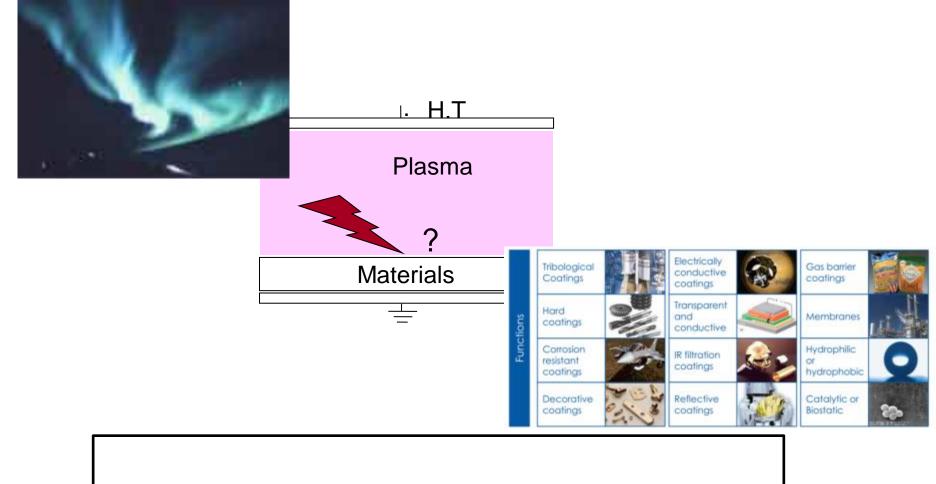
• deposition processes for semiconductor thin films for solar cells (α -Si:H)

90's

•Polymers, textiles, packaging, biomaterials, paper, ceramics, metals, MEMS, composites, etc. fully exploit plasmas

Electrically Tribological Gas barrier conductive Coatings coatings coatings Transparent Hard and Membranes Functions coatings conductive Hydrophilic Corrosion IR filtration resistant coatings coatings hydrophobic Reflective Catalytic or Decorative Biostatic coatings coatings

Consider PLASMA a tool to engineer materials

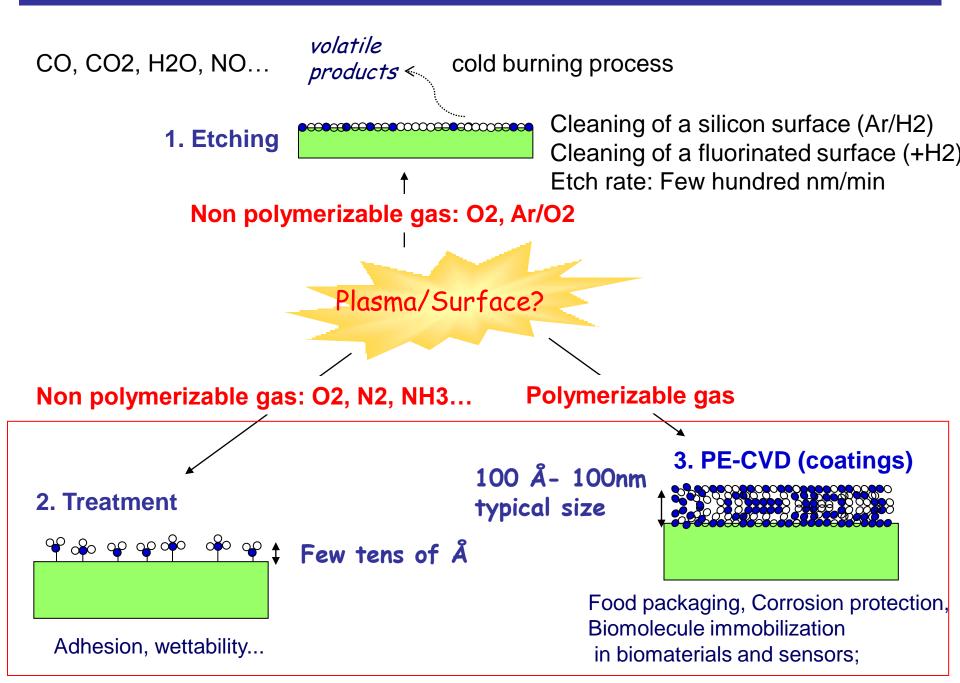


Plasma/surface Interaction ?

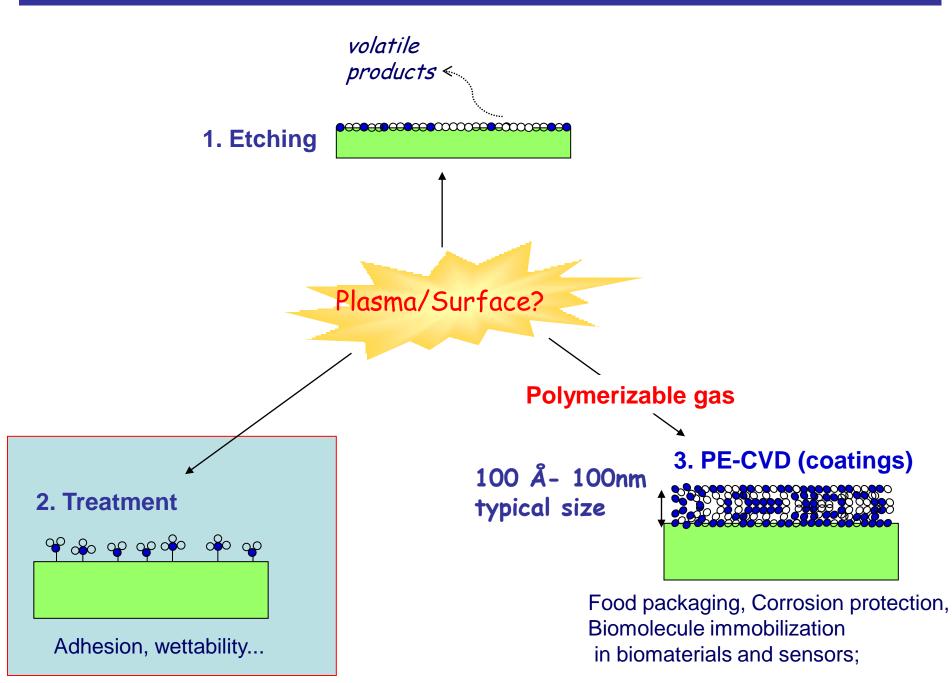
Plasma processing of materials?

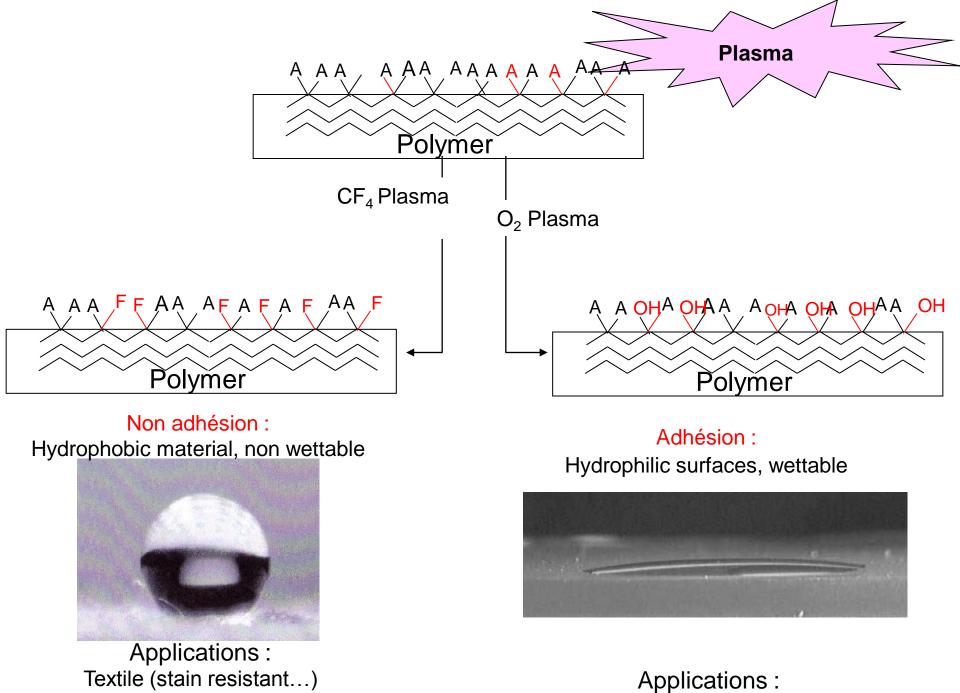
Positive and negative effects of plasma processing?

SURFACE MODIFICATION PROCESSES IN COLD, LOW PRESSURE PLASMAS



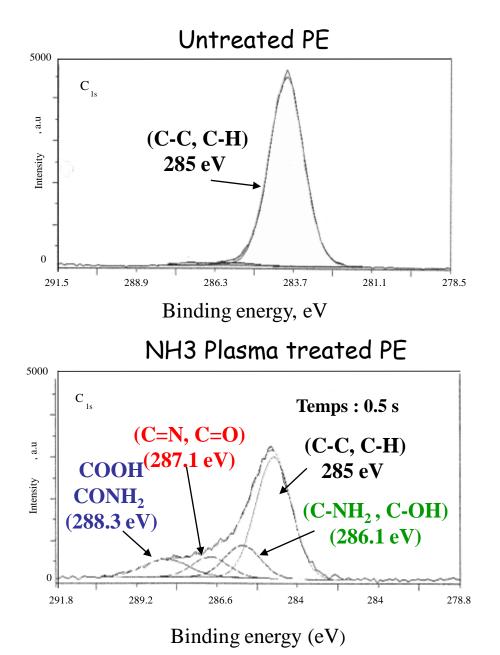
SURFACE MODIFICATION PROCESSES IN COLD, LOW PRESSURE PLASMAS



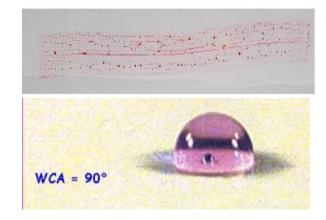


metal/polymer adhesion , ink/polymer

Plasma surface modifications of PE with NH3 plasma

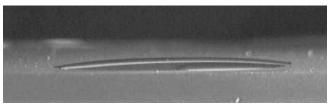


Untreated PE



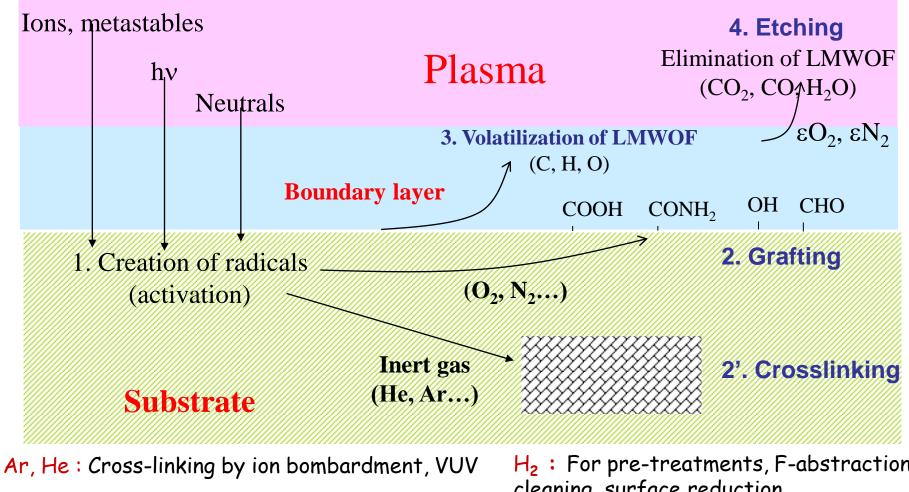
NH3 Plasma treated PE





Adhesion

PLASMA TREATMENTS OF POLYMERS



O₂, air, CO₂ ...: Increased surf. energy & wettability, sterilization, etching occurs.

 $CF_4, CF_4/O_2 \dots$

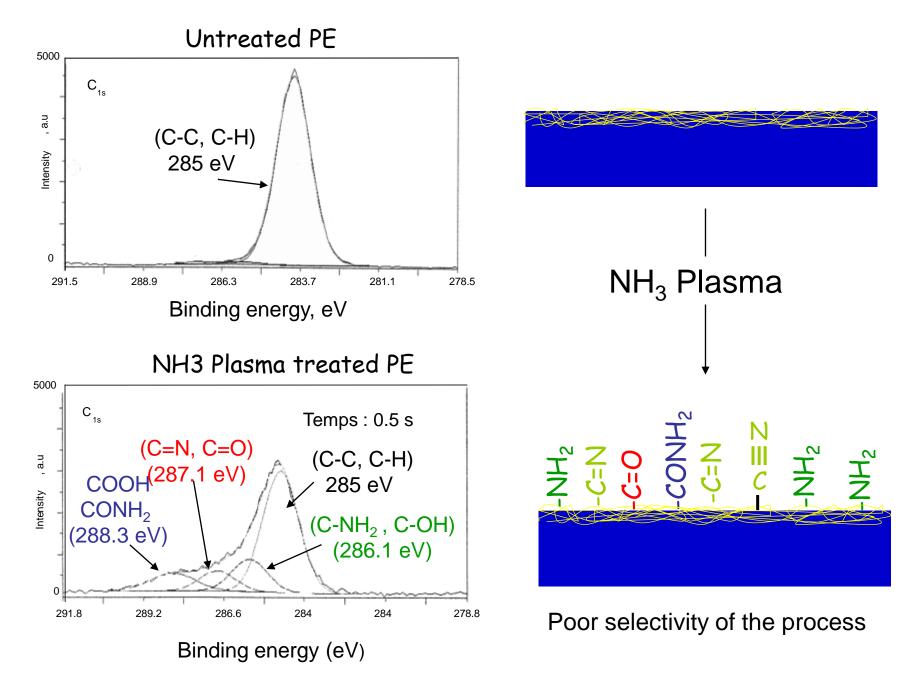
etching occurs,

Textile (stain resistant...)

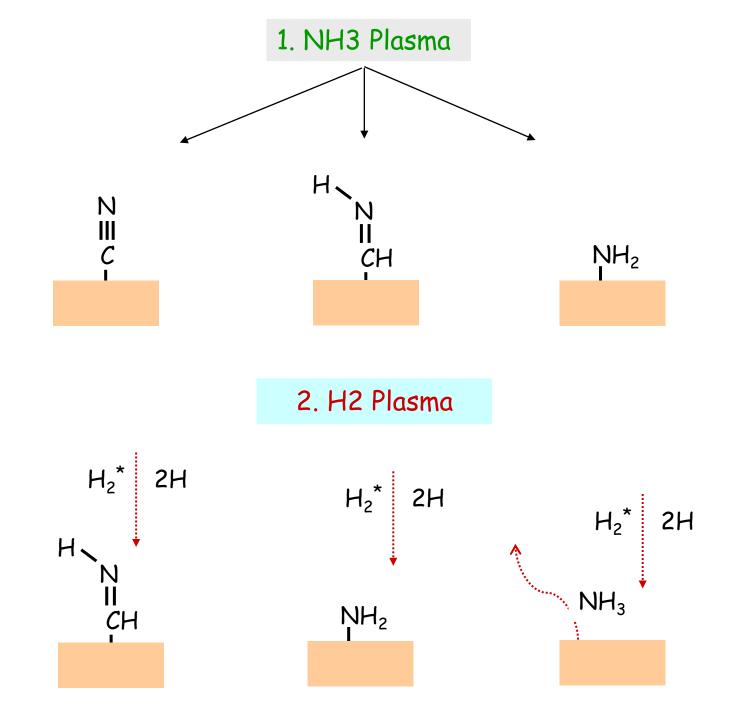
H₂: For pre-treatments, F-abstraction, cleaning, surface reduction. Limited polymer degradation, mild treatments.

N₂, NH₃ ...Grafting N-groups, increased surf. energy & wettability, limited etching, poor selectivity, sensitive to ageing Drawback 1: Poor selectivity of the plasma treatment process

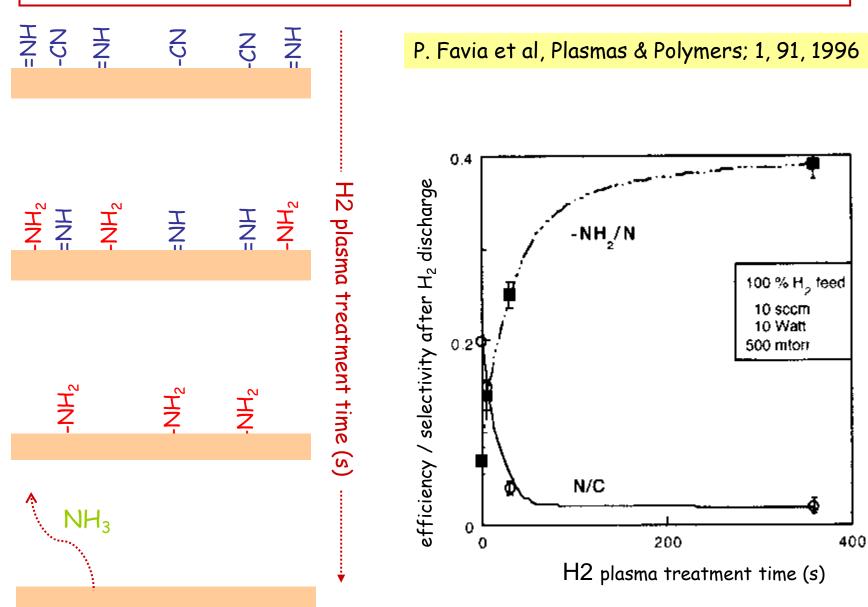
Drawback 1: Poor selectivity of the plasma treatment process



SOLUTION???

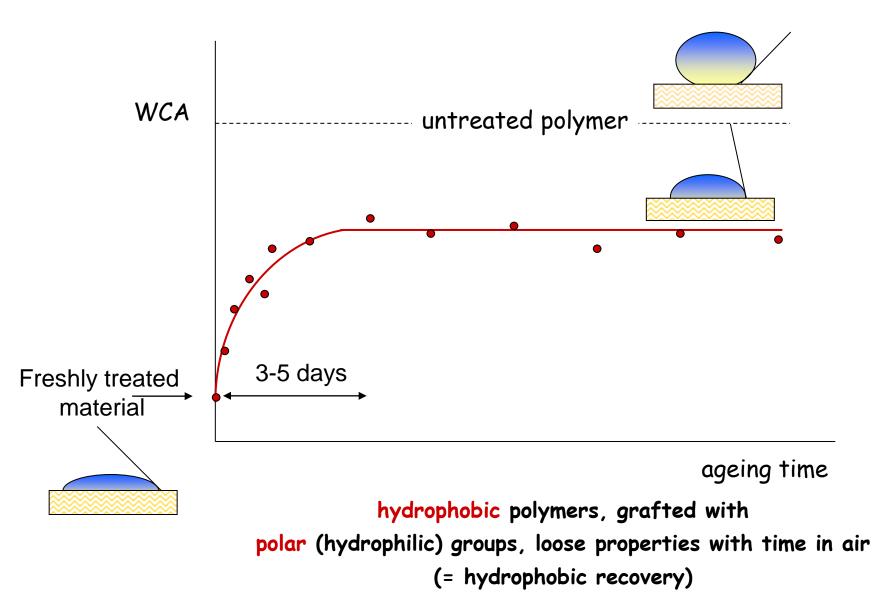


Selective grafting of $-NH_2$ groups on PE by means of plasma treatments in NH_3 and H2

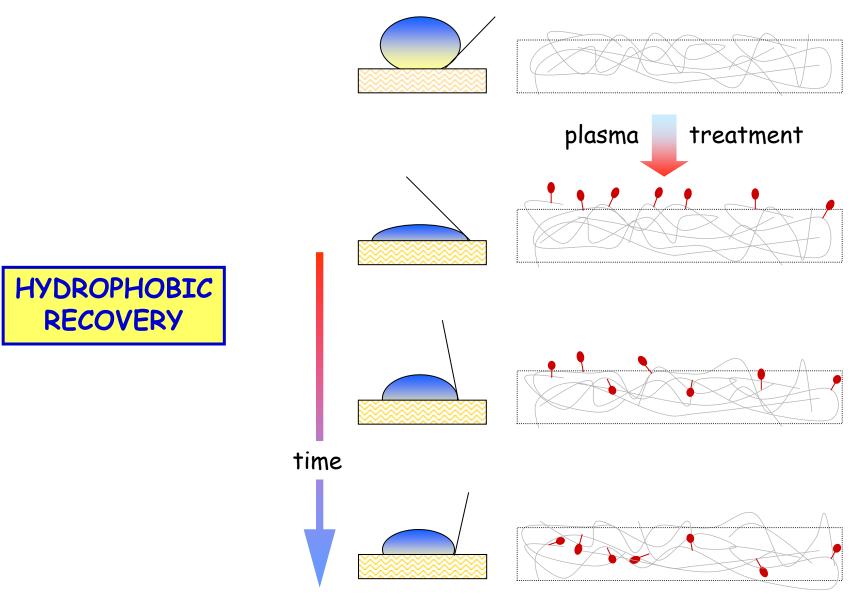


Drawback 2: Plasma treated polymeric materials aged quickly...

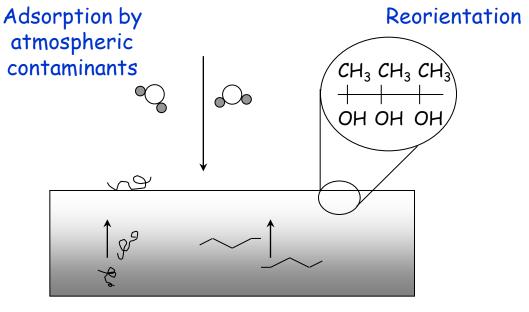
Ageing of plasma treated materials



The physical problem = ageing



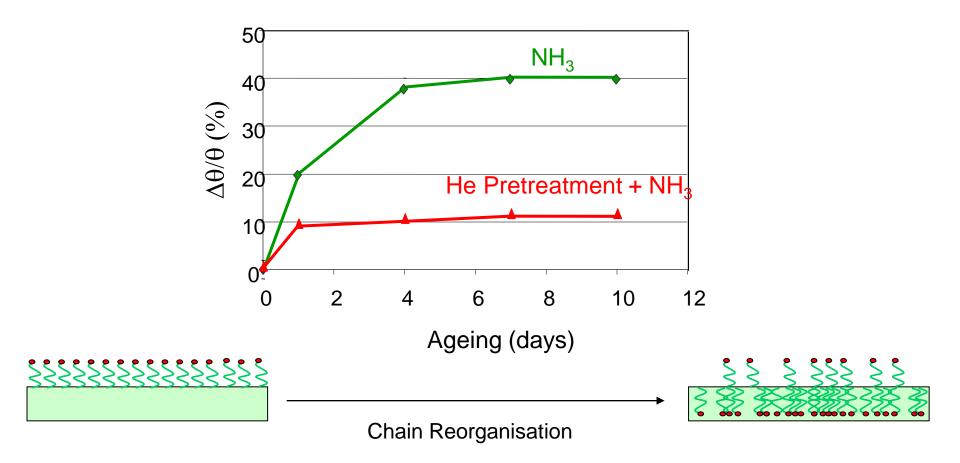
Mécanisms



Diffusion of additives or low molecular weight fragments

To avoid superficial degradation of the topmost surface layer To reinforce the cohesive strength of the tompost surface layer via CASING process

How to limit ageing?

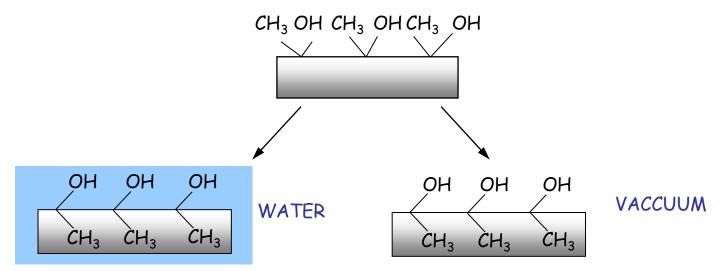


To limit the chain mobility by crosslinking reactions

Other solution??

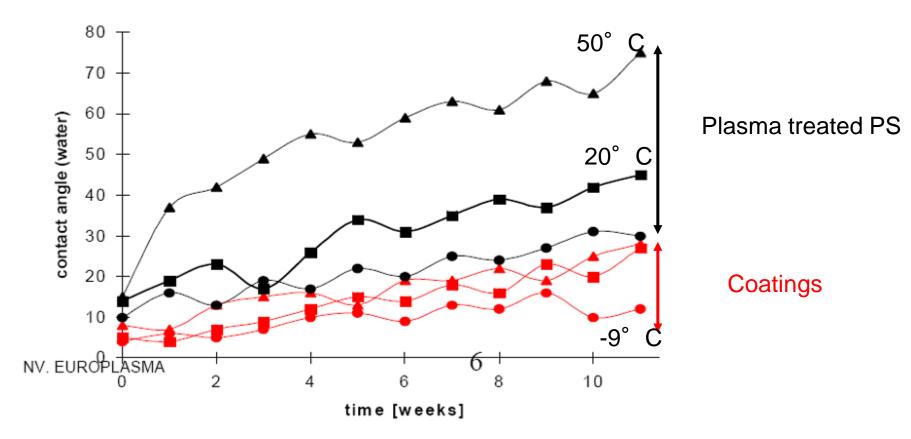
How to decrease the ageing of PT polymers?

- 1. Immobilization of adherent immediately after the treatment
- 2. To avoid the rearangement of grafted polar groups



3. To crosslink the topmost surface layer by He pretreatment

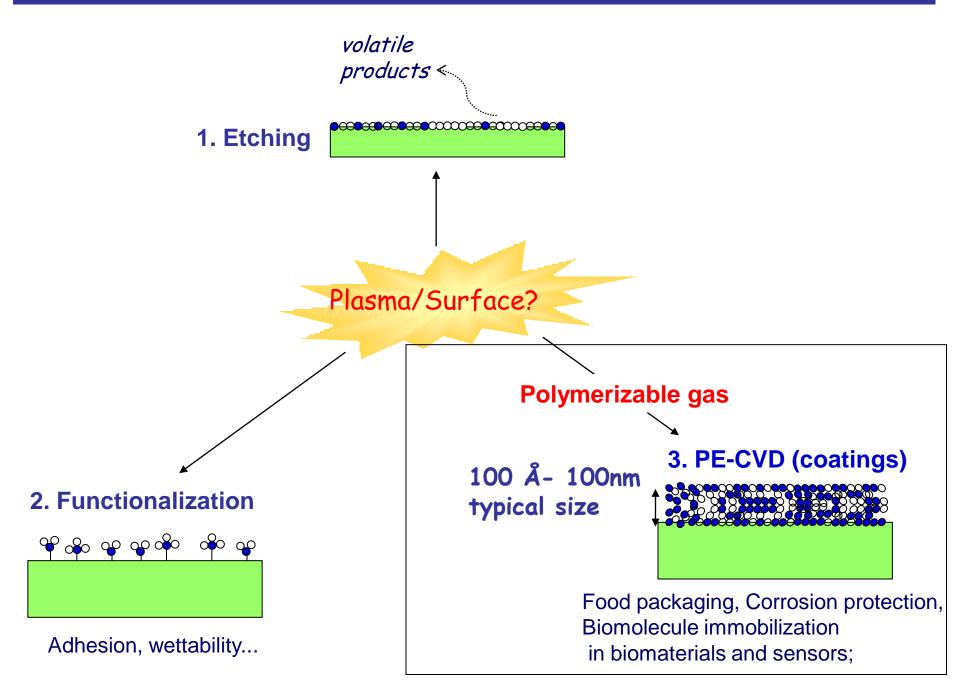
Other solution \rightarrow Plasma deposition of a thin coating



Aging of a plasmatreated PS-surface

EUROPLASMA

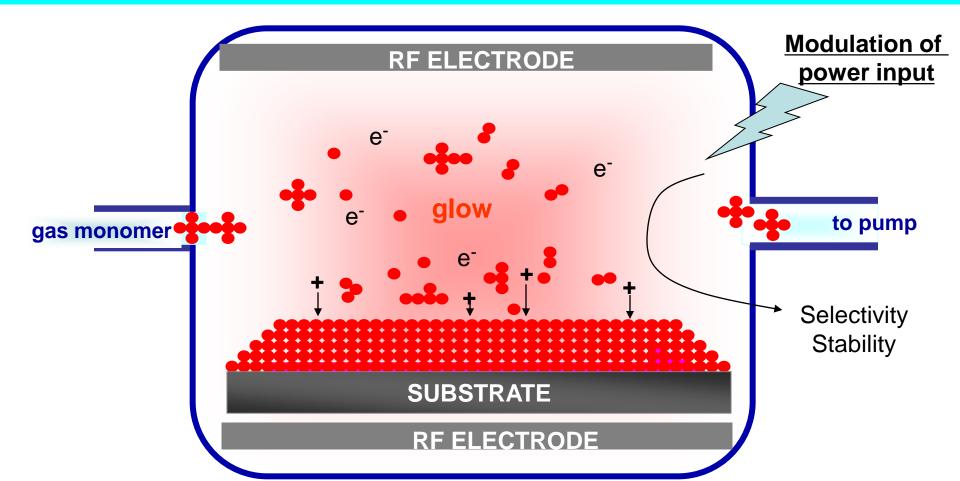
SURFACE MODIFICATION PROCESSES IN COLD, LOW PRESSURE PLASMAS



2. Polymerisable gas \rightarrow PECVD

PE-CVD PLASMA ENHANCED CHEMICAL VAPOR DEPOSITION

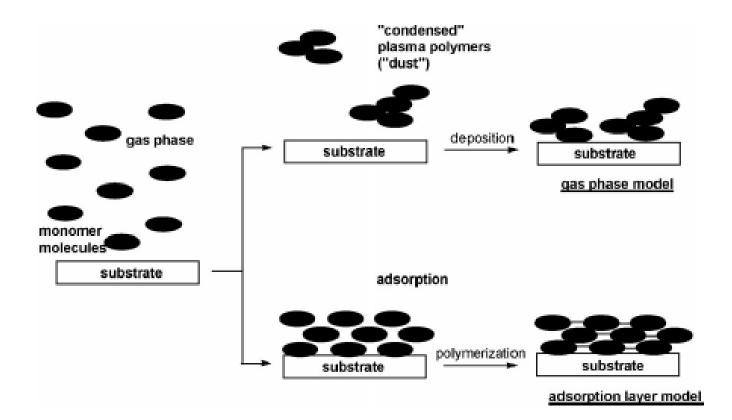
Inorganic (SiO₂, DLCs...) and organic (silicone-, PEO- teflon-like...) coatings can be deposited. PLASMA POLYMERIZATION is jargon name for PE-CVD of organic coatings;



Mechanisms of Plasma Polymerization?

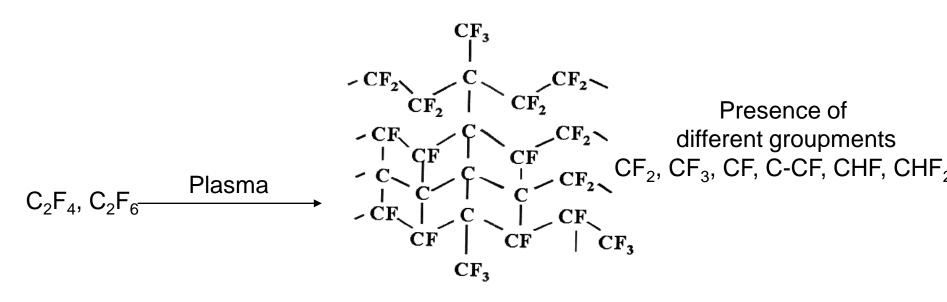
The preferred locus of plasma polymerization is a subject of controversy...

(i)In the gas phase (ii)Or in the layer adsorbed on the substrate



Jörg Friedrich, Plasma Process. Polym, 2011, 8, 783-802

Plasma polymerisation process



3D structure (depending on the degree of crosslinking)

Specificity of plasma polymerised coatings

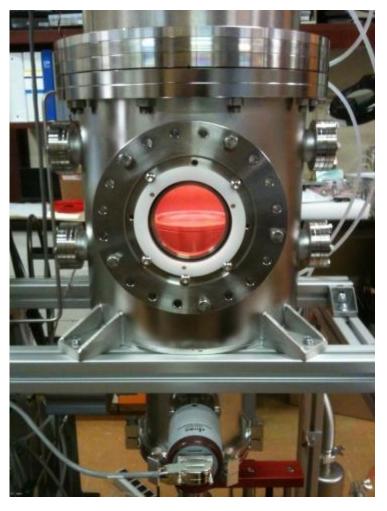
- 1. Ultra thin, pinhole-free layers of various compositions can be deposited as plasma polymers
- 2. Not a conventional polymerisation process: No repetition of the structure of the initial monomer

1. The physico-chemical properties of the coating are controlled by the plasma parameters (power, pressure) and not only by the starting monomer

Plasma-enhanced chemical Vapor deposition Reactor for Organic/inorganic coatings







Today:

there is a strong need to retain the momomer structure during plasma polymerisation process

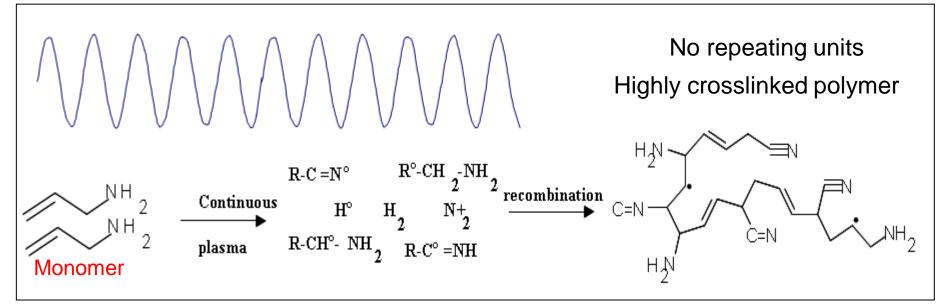
X : Functional group= NH2, COOH, SH...

To control the reaction selectivity To control the stability of the coating

Use of modulated discharges (Pulsed discharges)

How to control the selectivity of the process?

Polymerization under continuous wave plasma



H. Yasuda introduced an important factor (Yasuda factor) YF = W/FM (wattage/Molar mass x monomer flow)

For the case of high values of W/FM, all monomer molecules are extensively fragmented into single atomsand the monomer composition and –structure is barely recognizable in the resulting plasma polymer structure and composition

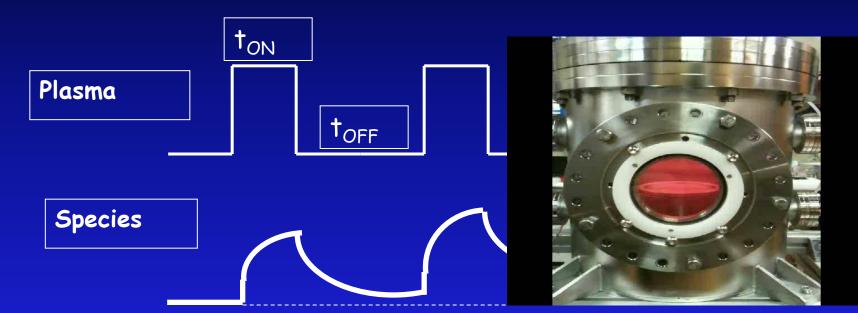
 $ABCDEF + plasma \rightarrow A + B + CD + E + F \rightarrow [FCABDE]n$

Yasuda proposed a new mechanistic concept in the 70's which he called « Atomic polymerization »

How to use LOW values of W/FM ??

Introduction of the pulsed-plasma technique, which corresponds to the greatest Innovation in the field of plasma polymerization was ...

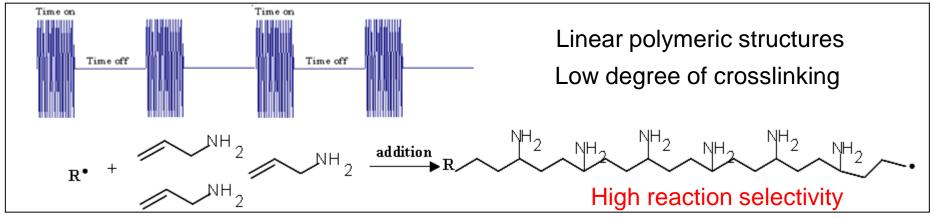
POWER MODULATION



Modulation Parameters

Period = t_{on} + t_{oFF} Duty Cycle = (t_{on}/Period)*100 Effective power $W_{eff} = W_{tot} \times DC$

Polymerization under pulsed discharge



A clever choice of « monomer » must be done to produce specifc charcateristics

corrosion resistant coatings

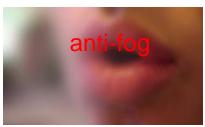
(Silica coatings from Organosilanes...)

transparent barrier Films for food packaging





antifouling coatings Ether rich coatings From glycols precursors



poly(vinyl alcohol) (PVA) layers

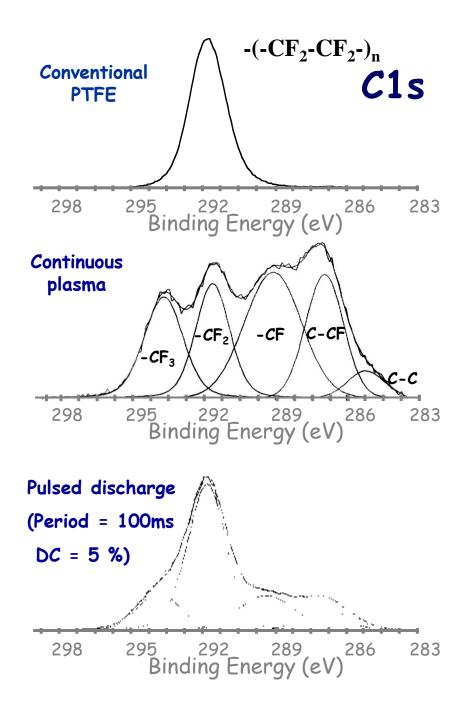


Scratch and abrasion-resistance (DLC coatings from CH4/Ar plasmas)



Teflon like coatings... from fluorinated monomers anti-ice on plane wings

Super hydrophobic coatings



Decrease of the DC (low W_{eff})

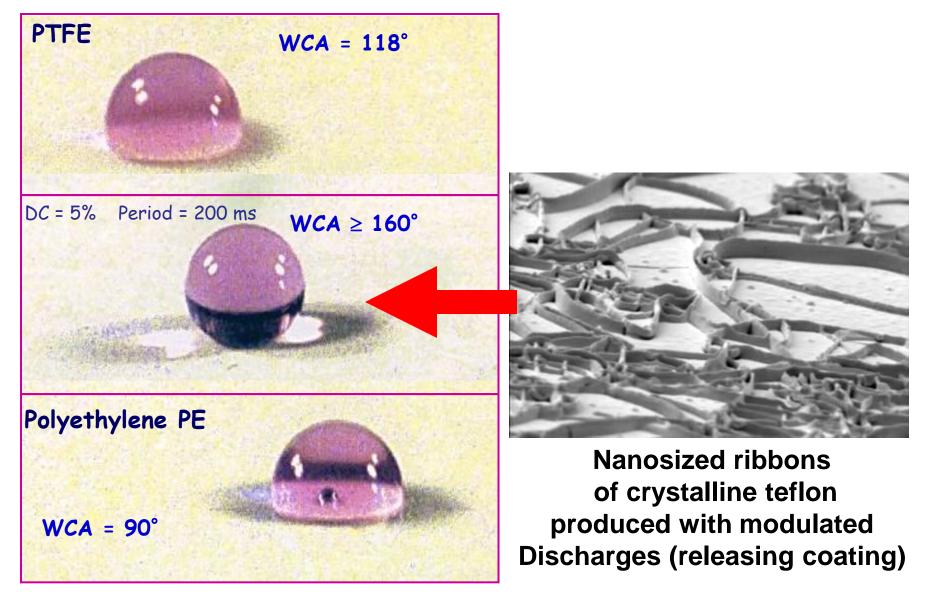




"teflon-like" structure

R. d' Agostino (Bari, Italie) Plasmas Processes and Polymers, 2004

"Super Hydrophobic" Plasma Deposited Coating



R. D'Agostino et al.

Plasma processes for cell adhesion

INHIBITION

PEO-like coatings

 $\mathsf{PEO} \quad -(\mathsf{CH}_2\mathsf{CH}_2\mathsf{O})_{\mathsf{n}}-$

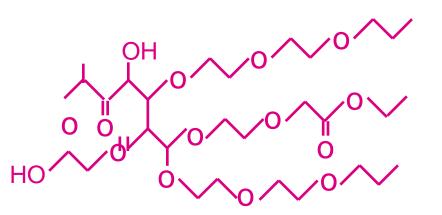
vot da-

PROMOTION

Grafted N-groups

-COOH functional coatings

PEO-like



Key Parameter retention of the PEO structure in the coating

Feed :glycols

H-tert BJ1 fibroblasts on PEO-like 5W, a cell-repulsive surface

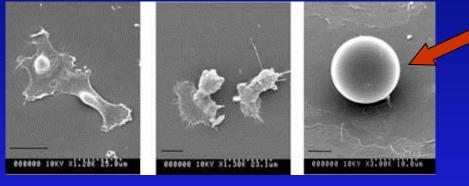
1. Oligoglymes ((CH_3 -O-(CH_2 - CH_2 -O)_n- CH_3 , n=1-4)

PEO - 5W

R. D' Agostino et al. Plasma processes and Polymers, Vol 1:57-62 (2004)



Diglyme (n=2) Di(ethylene glycol) dimethyl ether ou 2-Methoxyethyl ether ($C_6H_{14}O_3$)



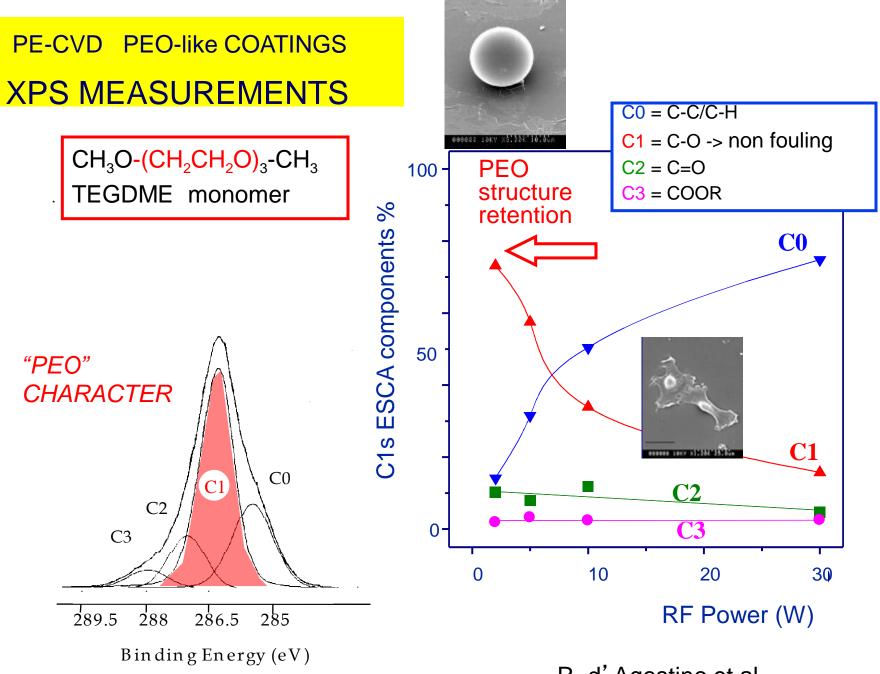
PEO -10W

DIGLYME 5W :

- antifouling +++ sur fibroblastes
- Surface de contact cellule/support minimale

PEO15W

R. Gristina et al.



R. d' Agostino et al.

TRANSPARENT BARRIER FILMS SIOX

featuring

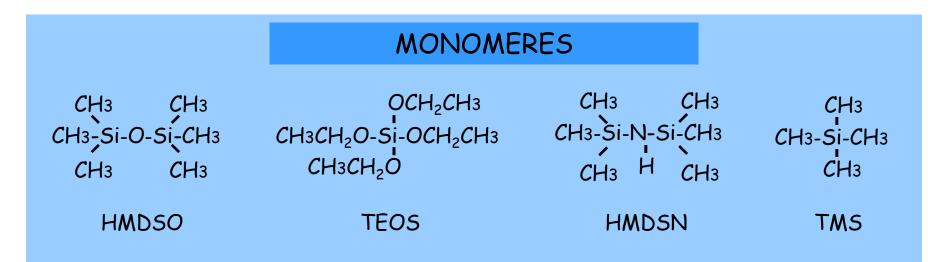
• LOW GAS TRANSMISSION RATE (food, pharmaceutical packaging)

- MW compatibility
- HARDNESS
- TRANSPARENCY
- INERTNESS
- · CORROSION RESISTANCE
- DIELECTRIC PROPERTIES

•in optimum conditions, O_2 GTR in industrial scale ranging between 3 and 10 cm³ m⁻² day

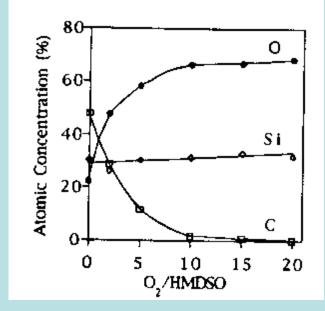
Key Parameters

monomer/ O_2 ratio input power (fragmentation)

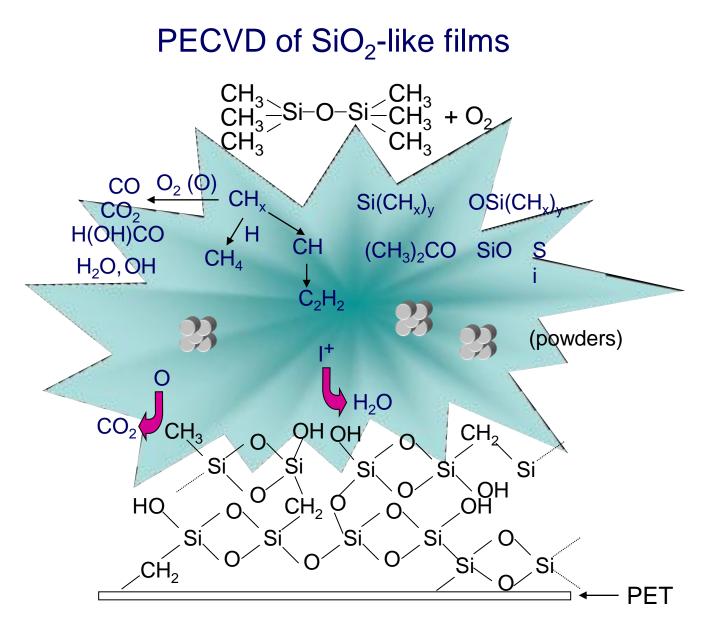


HMDSO/O2 RF GLOW DISCHARGES

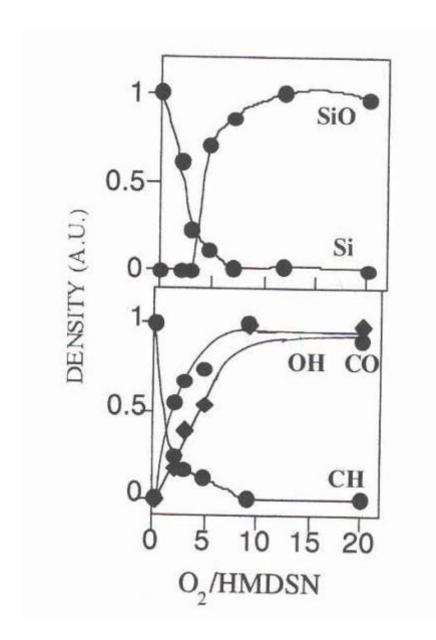
SiOx film composition at different feed ratios



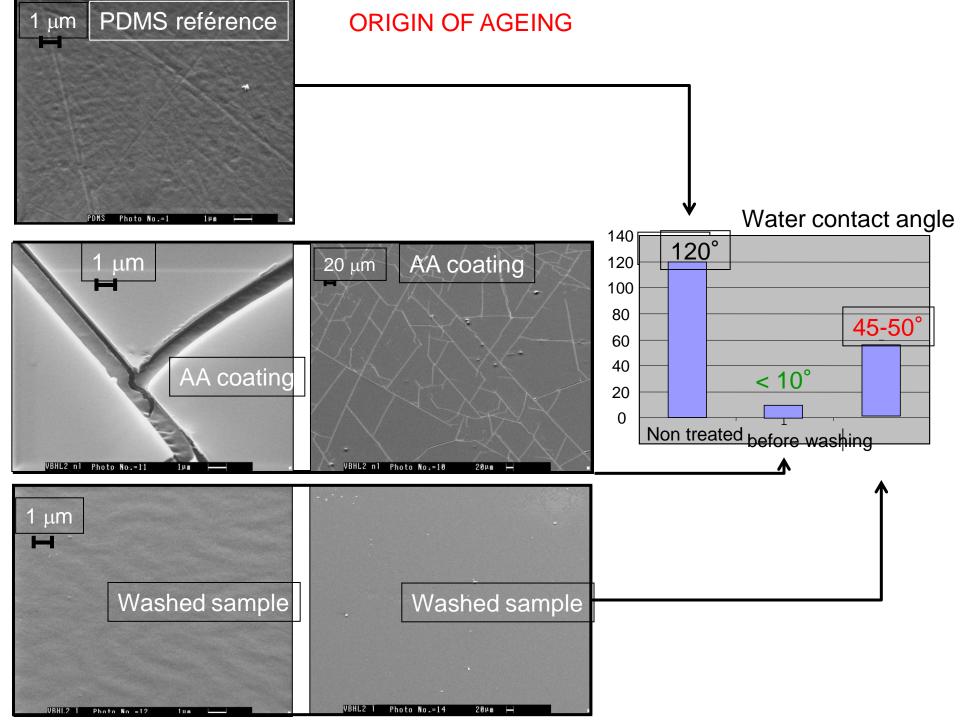
R. D'agostino et al.



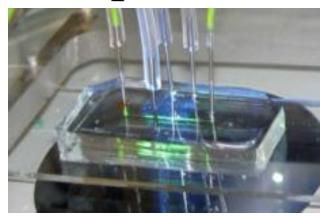
Silanols drill holes in high density SiOx

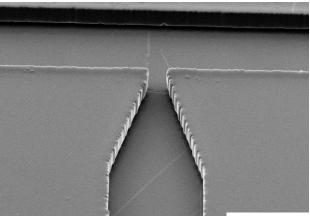


Stability??



Use of microfluidic devices to understand the properties of plasma deposited materials





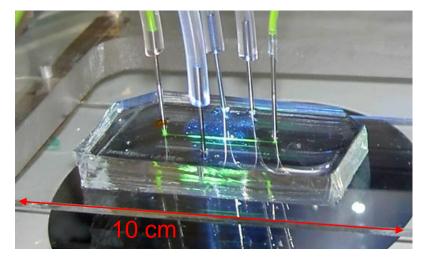
IPGGM: Pierre Gilles de Gennes Institute for microfluidic

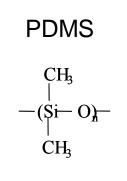
INSTITUT PIERRE-GILLES DE GENNES Pour la microfluidique

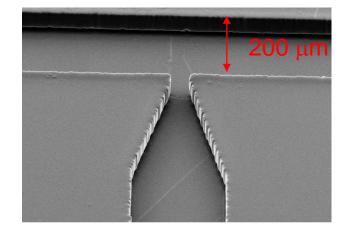




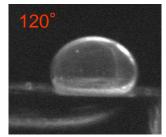
Microfluidic device : A tool to evaluate the stability of Plasma polymerized coatings

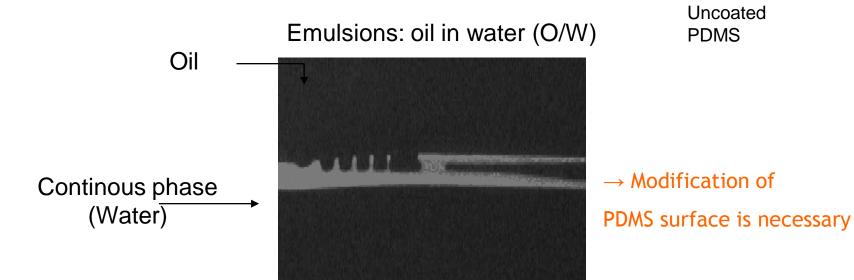




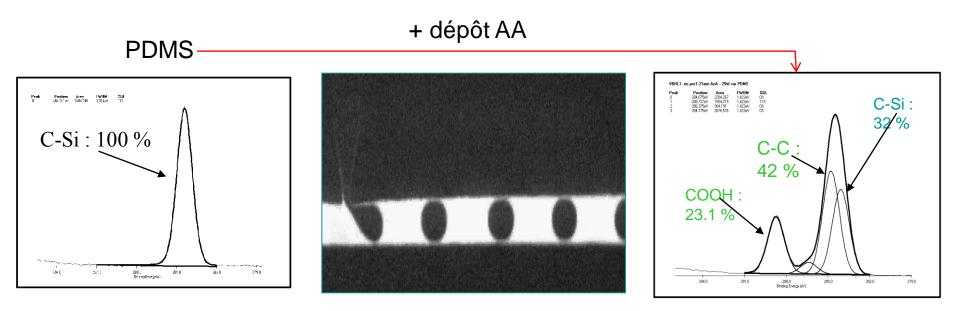


- Elastomer, rather inert, hydrophobic
- Dedicated material for micro-fabrication

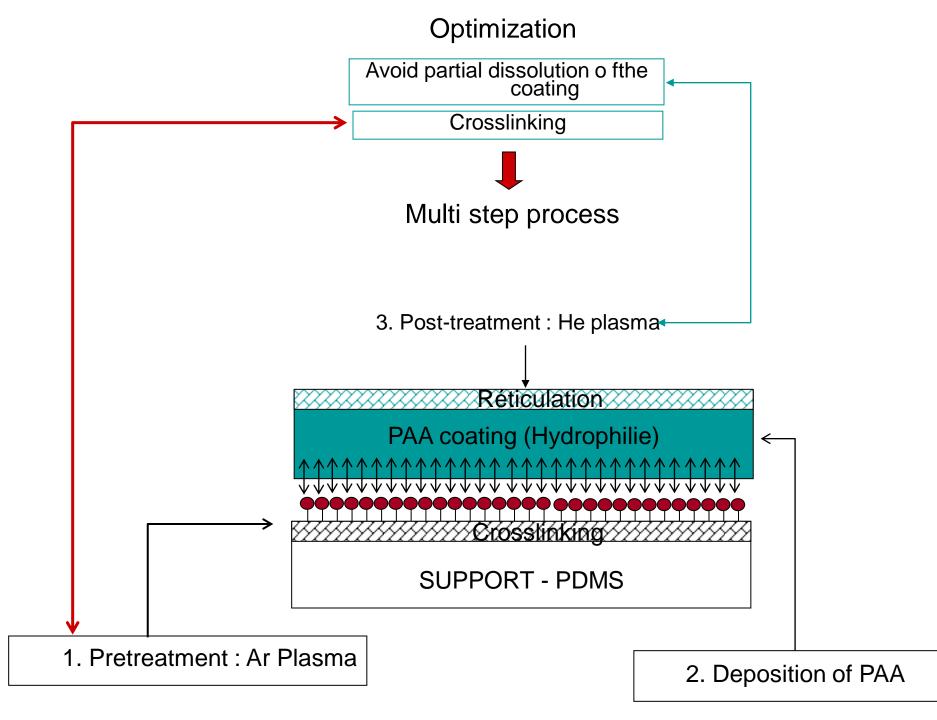


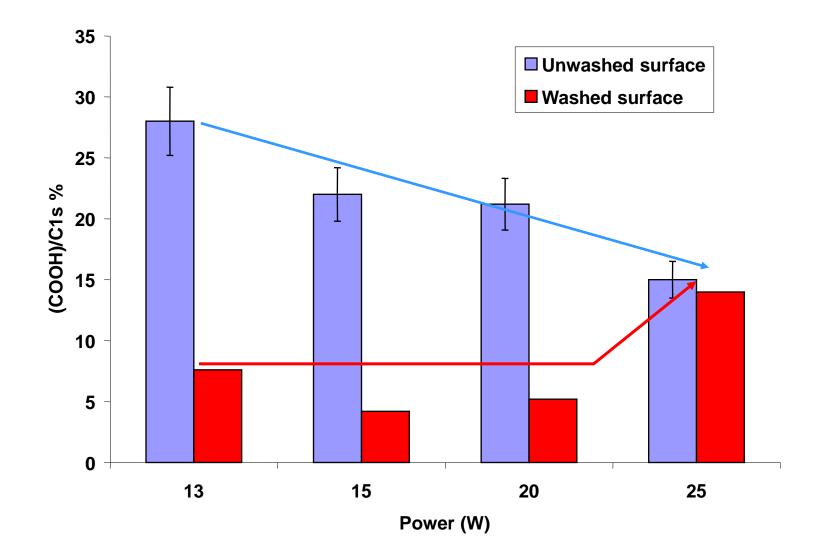


After optimization of the plasma process parameters...



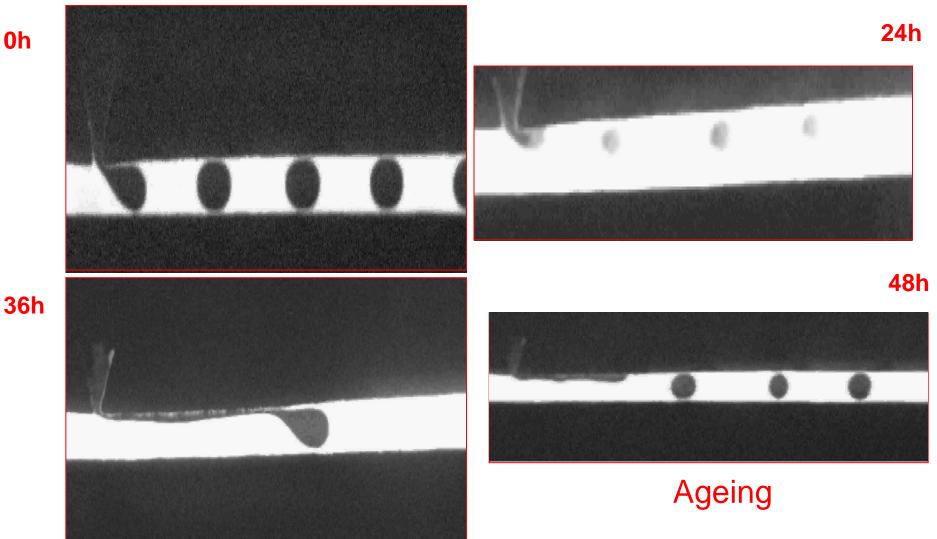
wettable





R. Jafari, et al., Thin Solid Films (2009), doi:10.1016/j.tsf.2009.03.217

Use of an O2 plasma treatment

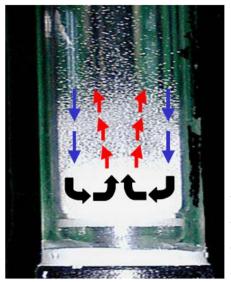


0h

Treatment of flat substrates is easy... 3D objects?

Plasma treatment of powders in a Fluidized Bed Plasma Reactor Diameter: few microns to hundred of microns...









- Continuous agitation of the particles: homogeneity in the treatment of powders
 Decrease of the particle agglomeration
- ✓ Large exchange surface between the particles and the gas phase (plasma)
 - Excellent heat transfer: reducing the risk of thermal degradation

Industrial plasma reactors...

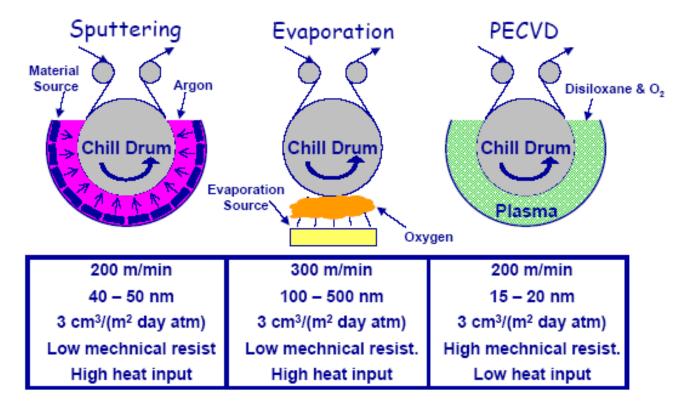


With the courtesy of Prof. D' Agostino

Dépôts de couches SiO_x (couches barrière) sur films de PET pour les applications alimentaire

Research & Development 💮 🕐 One World 關 One Team 💿 One Vision

Industrial Coating Technologies for SiOx Barrier Material

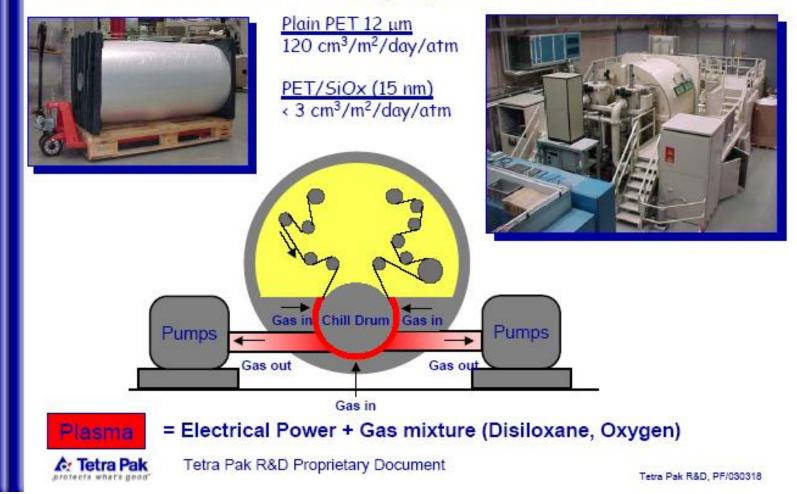


Tetra Pak R&D Proprietary Document

A: Tetra Pak

Research & Development 💮 One World 🗰 One Team ④ One Vision

Industrial PECVD Equipment at Tetra Pak



Our plasma reactors...



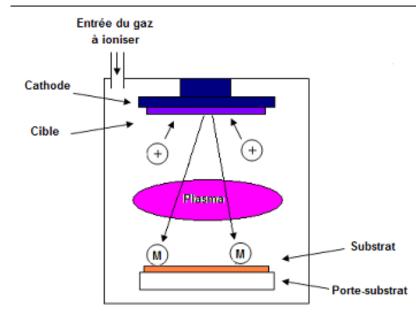
PECVD/Sputtering/Evaporation



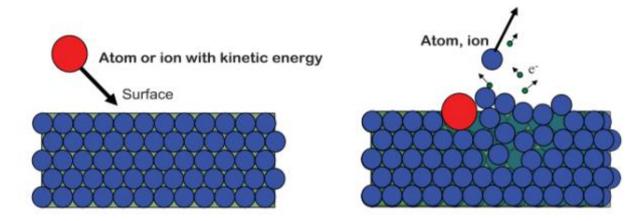




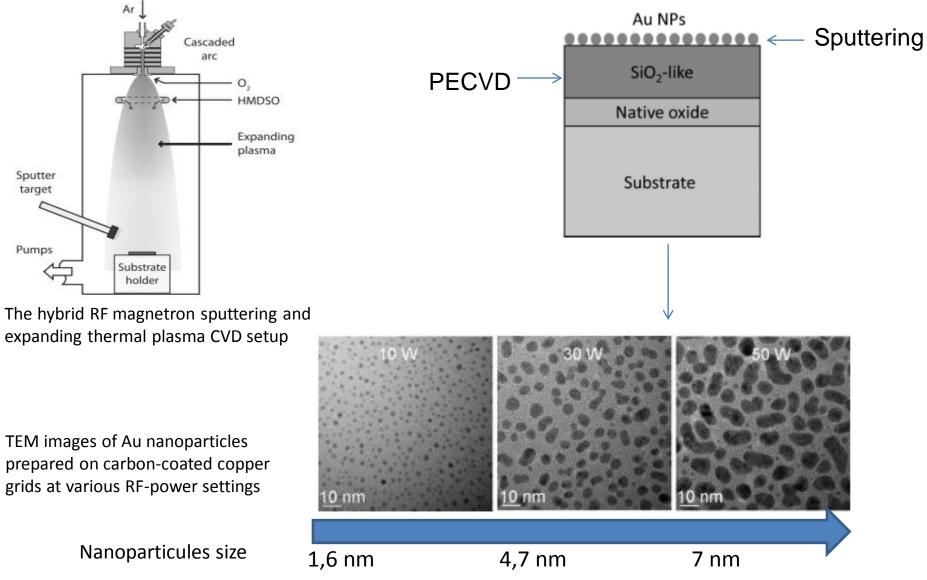
Other technique : RF SPUTTERING TECHNIQUE







HYBRID SPUTTERING-REMOTE PECVD: Deposition of Au Nanoparticles on SiO₂ Layers

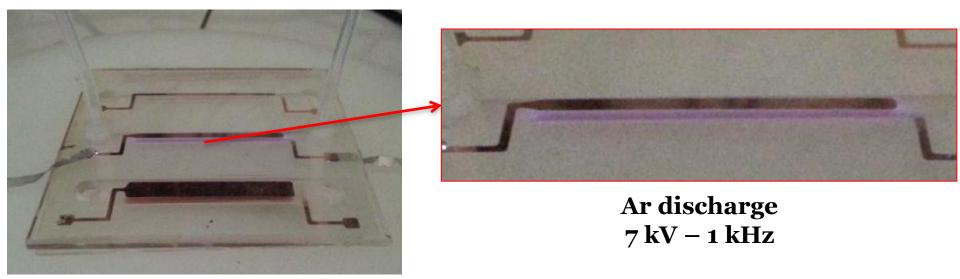


Haile Takele Beyene, Frans D. Tichelaar, Paul Peeters, Ivan Kolev, Mauritius C. M. van de Sanden, Mariadriana Creatore. Plasma Process. Polym. 2010, 7, 657–664





µChannel : Width = 4 mm Length = 50 mm Depth = 500 μm



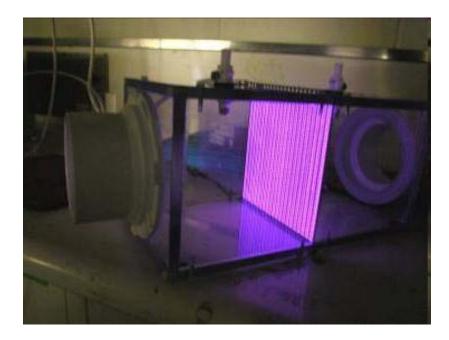
Other applications...

Non-thermal plasma in air at atmospheric pressure

Typical applications:

- Ozone production
- -Treatment of polluted gaseous effluents

DBD device for air treatment



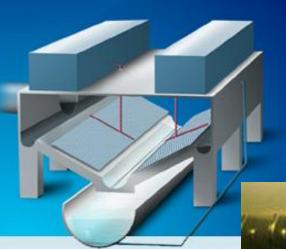


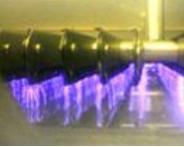


our technology

The expanding pollution of drinking water sources and increasingly stringent regulations promote the development of new, innovative treatment technologies.

Conventional water treatment methods are often not effective in the removal of micro-pollutants such as Volatile Organic compounds (VOC's) because of their physical and chemical characteristics.





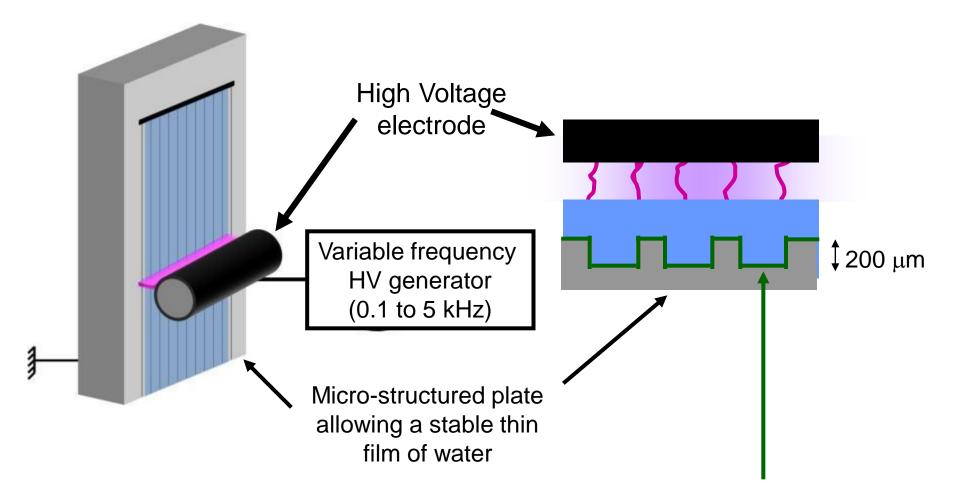
Henry's law constants at 25° C (atm.m³.mol-1)

kW/hr/m3 kW/hr/1000gal

тсе	1.21
MIB	1.67
NDMA	0.63
1,4 dioxane	1.24
мтве	3.8

heptanol	2×10 ⁻⁵
TCE	1×10 ⁻⁴
MIB	6.5×10 ⁻⁵
NDMA	3.34×10 ⁻⁵
1,4 dioxane	5×10 ⁻⁵
MTBE	6 ×10 ⁻⁴

New non-thermal plasma reactor

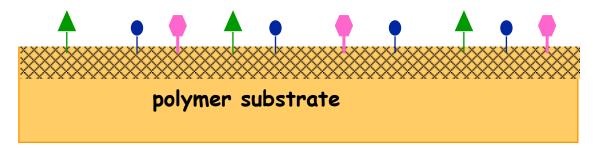


The micro-structured electrode is covered by a coating exhibiting catalytic and hydrophilic properties

CONCLUSION : SURFACE MODIFICATION OF MATERIALS WITH LP PLASMA TECHNIQUES

PLASMA TREATMENTS: Modification of the topmost layers of materials (polymers) by grafting chemical groups

Selectivity : (-) \rightarrow Controlled by gas feed Stability : Problem of ageing \rightarrow Additional pretreatment in inert gas (He, Ar...)



PE-CVD PLASMA ENHANCED CHEMICAL VAPOR DEPOSITION

High reaction selectivity : (+) \rightarrow Controlled by the modulation of power input Stability : Problem of dissolution \rightarrow Control of the degree of crosslinking

CONCLUSION : Consider PLASMA a tool for new applications

- Polymer activation for enhancing metal adhesion
- Polimer activation for enhancing color adhesion (e.g. car bumpers, plastic bags, textiles, ...)
- Stain-resistant clothings and garments
- Super-hydrophobic coatings
- Corrosion resistant coatings for alloies



- Tissue engineering and microstructuring of polymers for contact guidance of cells
- Anti-trombotic coatings for prostheses
- Bacterial resistant materials for food packaging and prostheses
- \cdot Non fouling coatings

Acknowledgements

Partnership & Collaborators







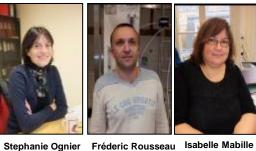






MINISTÈRE DE L'ENSEIGNEMENT SUPÈRIEUR ET DE LA RECHERCHE

















Daniel Bonn Willy Morscheidt



Dalila Chili Daniela D'Elia





Olivier Lesage



Rao Xi

Thank you for your

attention



Institut de Recherche de Chimie Paris



Alexandre MA



Bruno Pelat

Fatiha Laurence





Rafik Benrabbah Maxime Cloutier Diane Gumuchian

Guillaume Schelcher

Christos Anggelopoulos